

Lewis Diagrams practise

QUESTION: Draw the Lewis structure (electron dot diagram) for each molecule

PH ₃	CO ₂	H ₂ CO	OCl ₂ aka Cl ₂ O
O ₂	CH ₃ Br or CH ₃ Cl	SO ₂	SiCl ₄
H ₂ O	CH ₂ Br ₂ or CH ₂ Cl ₂	CS ₂	HCN
XO ₂ (where X has 4 valence electrons)	YO ₂ (where Y has 6 valence electrons)	NCl ₃	SF ₂
PBr ₃ or PCl ₃	H ₂ S	COCl ₂	F ₂ O